GL Sciences Inc.

Analysis of Formaldehyde in Drinking Water by **HPLC and Post-column Derivatization**

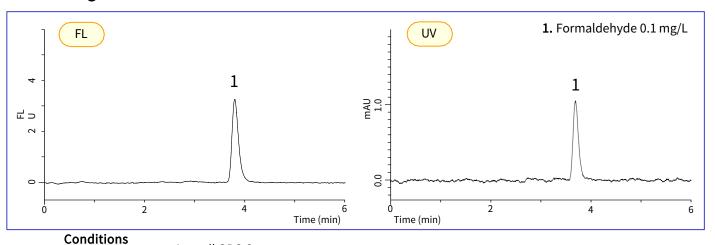
Formaldehyde is widely known as a toxic chemical, and the guideline value of 0.9 mg/L for its content in drinking water has been established by WHO. This note describes a determination method for formaldehyde in beer using HPLC equipped with a post-column derivatization system.

Formaldehyde in injected solution is separated from other compounds in the ODS column, reacted with

acetylacetone solution, and converted to fluorescent 3,5diacetyl-1,4-dihydrolutidine.

Concentration of one-tenth of the guideline value of formaldehyde was determined with a UV/VIS detector in the HPLC system. Furthermore, florescence (FL) detector provided even higher sensitivity.

Chromatograms obtained from standard solution



: Inertsil ODS-3 Column

 $(5 \mu m, 250 \times 4.6 \text{ mm I.D.})$

Cat.No. 5020-01732

: 6 mM Na₂HPO₄ (pH 2.1, H₃PO₄) Eluent

: 1.0 mL/min Flow rate :20°C

Col. Temp.

Reaction reaget

: Acetylacetone solution*

Flow rate of reaction reagent

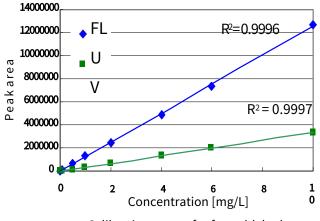
: 0.5 mL/min

: VIS 413nm Detection

FL Ex. 445 nm, Em. 505 nm

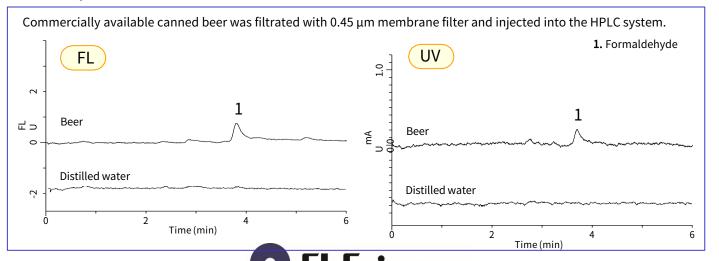
Injection vol. : 10 µL

To 150 g of ammonium acetate, 3 mL of acetic acid and 2 mL of acetylacetone was added. The solution was made up to 1 L with water.

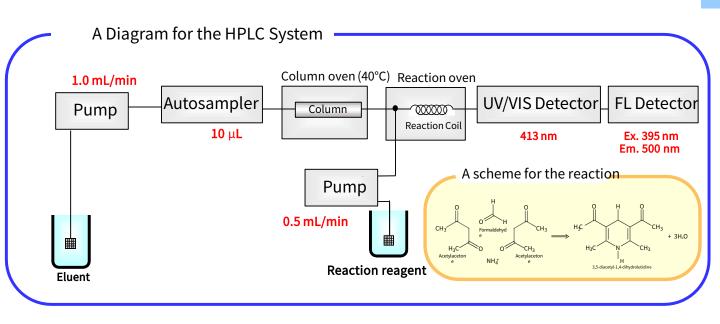


Calibration curves for formaldehyde

An analysis of canned beer



^{*} Acetylacetone solution:



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