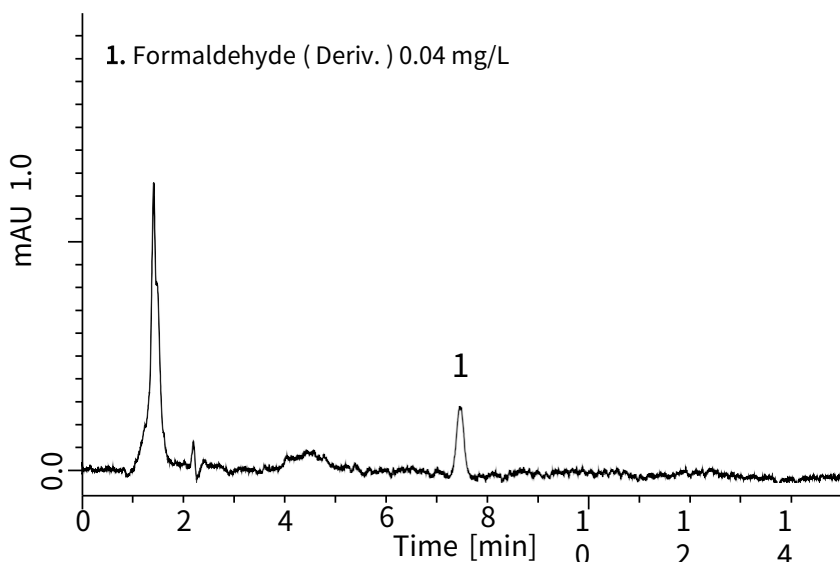


In this note, a determination method for formaldehyde is described using a GL-7400 HPLC system.

Formaldehyde is used as a component of various material, such as coating solution, adhesive agent, antiseptics, and resins. However, it is also a well-known chemical compound responsible for sick house syndrome. Formaldehyde released into indoor air can irritate eyes and mucous membranes. Furthermore, baby clothes containing formaldehyde cause severe skin inflammation.

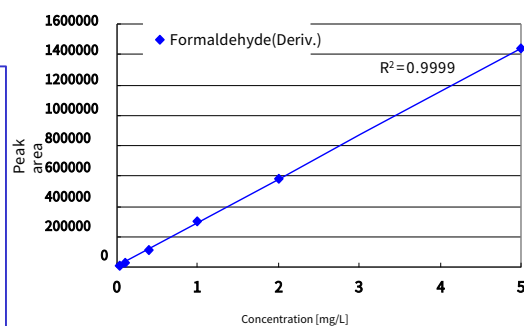
## A chromatogram obtained from standard solution

This is an example of an analysis of derivatized formaldehyde. The concentration described below is the concentration of the solution before derivatization. In case the pretreatment method in the next page is applied to a sample, the concentration used in this chromatogram 0.04 mg/L can be converted to 1.6 mg/kg.



Therefore, formaldehyde contents of household products are regulated by law in Japan. The contents of baby clothes must be less than 16 ppm (=mg/kg), while those of other household products are required to be less than 75 ppm.

In this note, the determination of formaldehyde in clothing was performed compliant with the enforcement regulations of the law. The water extracts from samples were derivatized by acetylacetone (2,4-pentandione) and injected to the HPLC system coupled with UV-VIS detector.



The calibration curve of formaldehyde

\* HPLC analysis was performed after the derivatization reaction of formaldehyde. The standard solution was gradually diluted and each were derivatized then analyzed. The concentration in the graph is the concentration of diluted solution before derivatizing using acetylacetone solution.

### Conditions

Column:	Inertsil ODS-3 (4 $\mu$ m, 150 x 4.6 mm I.D.)
Column Cat. No. :	5020-04645
Eluent:	A) CH <sub>3</sub> CN B) H <sub>2</sub> O A / B = 15 / 85, v/v
Flow Rate:	(gradient mixer)
Col. Temp.:	1.0 mL/min 40 °C
Detection:	VIS 413 nm
Injection vol.:	10 $\mu$ L

## sample pretreatment

### Sample

2.5g

### Extraction

Water 100 mL  
Place in a water bath at 40°C for 1 hour  
and shake a few times  
Filtration while warm

### Derivatization

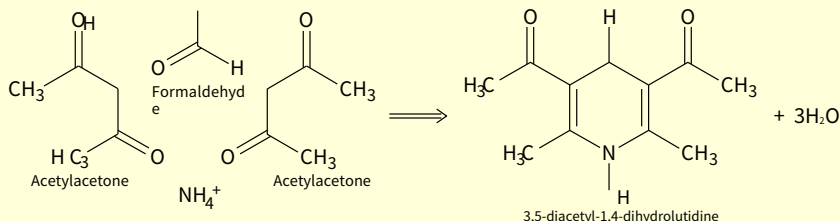
Extract 5 mL  
5 mL Acetylacetone solution\*<sup>1</sup>  
Place in a water bath at 40 °C for 30min  
Wait for 30 min

### HPLC-UV

\*1 acetylacetone solution:  
To 150 g of ammonium acetate, 3 mL of acetic acid and , 2 mL of  
acetylacetone were added. The solution was diluted with water and  
made up to 1 L.

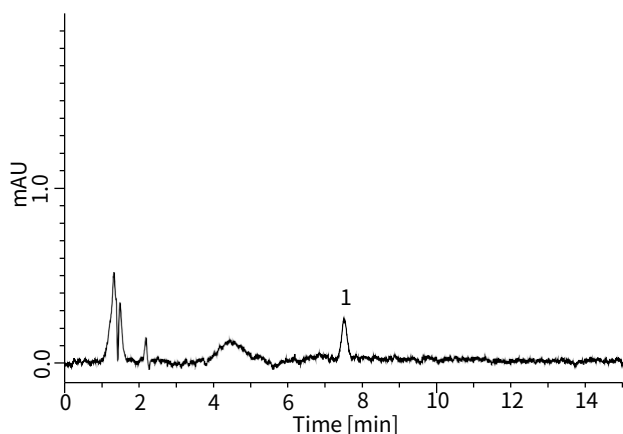
## Derivatization reaction

One molecule of formaldehyde reacts with two molecules of acetylacetone and one molecule of ammonium ion. As a result of the reaction, 3,5-diacetyl-1,4-dihydrolutidine, which can be sensitively detected by UV-Vis detector, is generated.



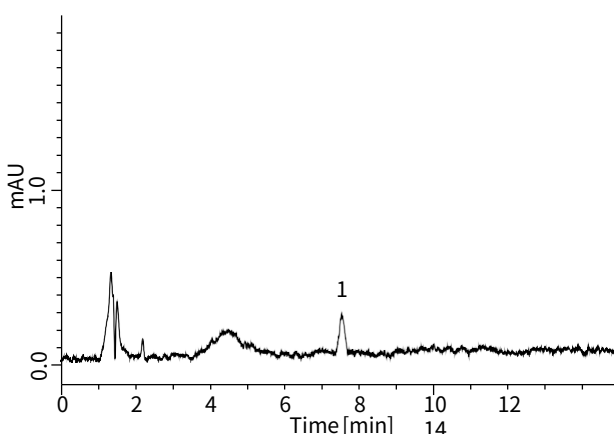
Structures are created using Chemistry 4-D Draw which is provided by ChemInnovayon Software, Inc.

## Polyester fiber extract



## Cotton extract

1. Formaldehyde (Deriv.)



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